

Enantioselective Synthesis of (3*R*,6*S*,7*R*,18*R*,19*S*)-, (3*R*,6*S*,7*R*,18*R*,19*R*)-, and (3*R*,6*S*,7*R*,18*S*,19*R*)-Quassiols A.

A Comment on the Stereochemistry of Natural Quassiol A

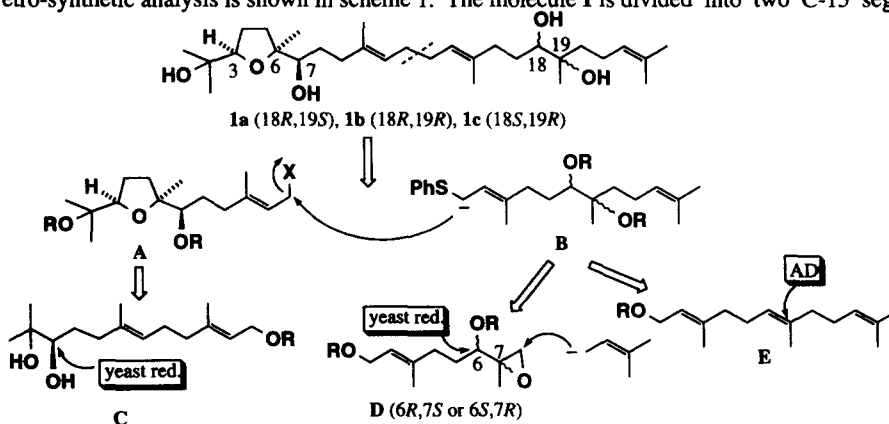
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Abstract: (3*R*,6*S*,7*R*,18*R*,19*S*)-, (3*R*,6*S*,7*R*,18*R*,19*R*)-, and (3*R*,6*S*,7*R*,18*S*,19*R*)-Quassiols A were synthesized enantioselectively by using baker's yeast reduction and asymmetric dihydroxylation, but their optical properties were not identical to that of natural quassiol A. Stereostructure of natural quassiol A was proposed.
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Quassiol A (**1**) is a triterpene ether isolated together with its 18-monoacetate, quassiol B (**2**), from *Quassia multiflora*.¹⁾ Although the plane structure has been determined on the basis of spectroscopic analyses, its stereostructure including absolute stereochemistry has not yet been clarified except for the relative configuration at C-3, C-6, and C-7 as shown. In order to determine the absolute stereostructure, we synthesized (3*R*,6*S*,7*R*,18*R*,19*S*)-**1a**, (3*R*,6*S*,7*R*,18*R*,19*R*)-**1b**, and (3*R*,6*S*,7*R*,18*S*,19*R*)-**1c**, enantioselectively. Although their optical properties were not consistent with those of natural product, we would like to propose the enantiomer of **1a** for the stereostructure of natural quassiol A.

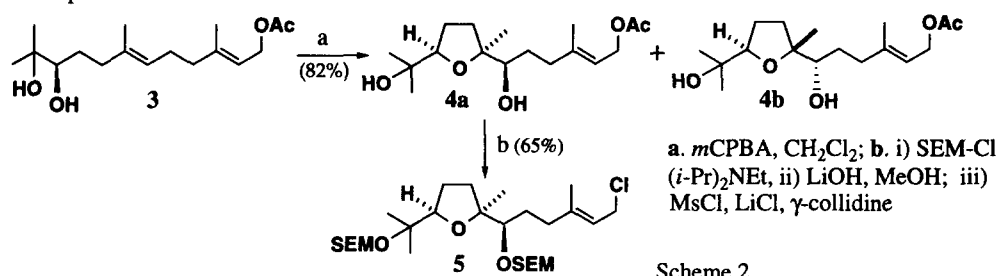
The retro-synthetic analysis is shown in scheme 1. The molecule **1** is divided into two C-15 segments A



Scheme 1

and **B** which would be derived from farnesol or geraniol derivatives -**C**, **D**, **E**- in enantioselective manner using baker's yeast reduction² and asymmetric dihydroxylation (AD)³ as the chirality induction method.

The synthesis of left-half segment **A** is illustrated in scheme 2. (10*R*)-10,11-Dihydroxyfarnesyl acetate (**3**) derived from farnesol in high enantiomeric purity² was treated with *m*-chloroperbenzoic acid to yield two diastereomeric tetrahydrofuran derivatives **4a** and **4b** in ca. 1 : 1 ratio. The stereochemistry was determined by applying modified Mosher's method^{4,5} unequivocally. The less polar (3*R*,6*S*,7*R*)-diol **4a** was then converted in three steps into a chloride **5**.



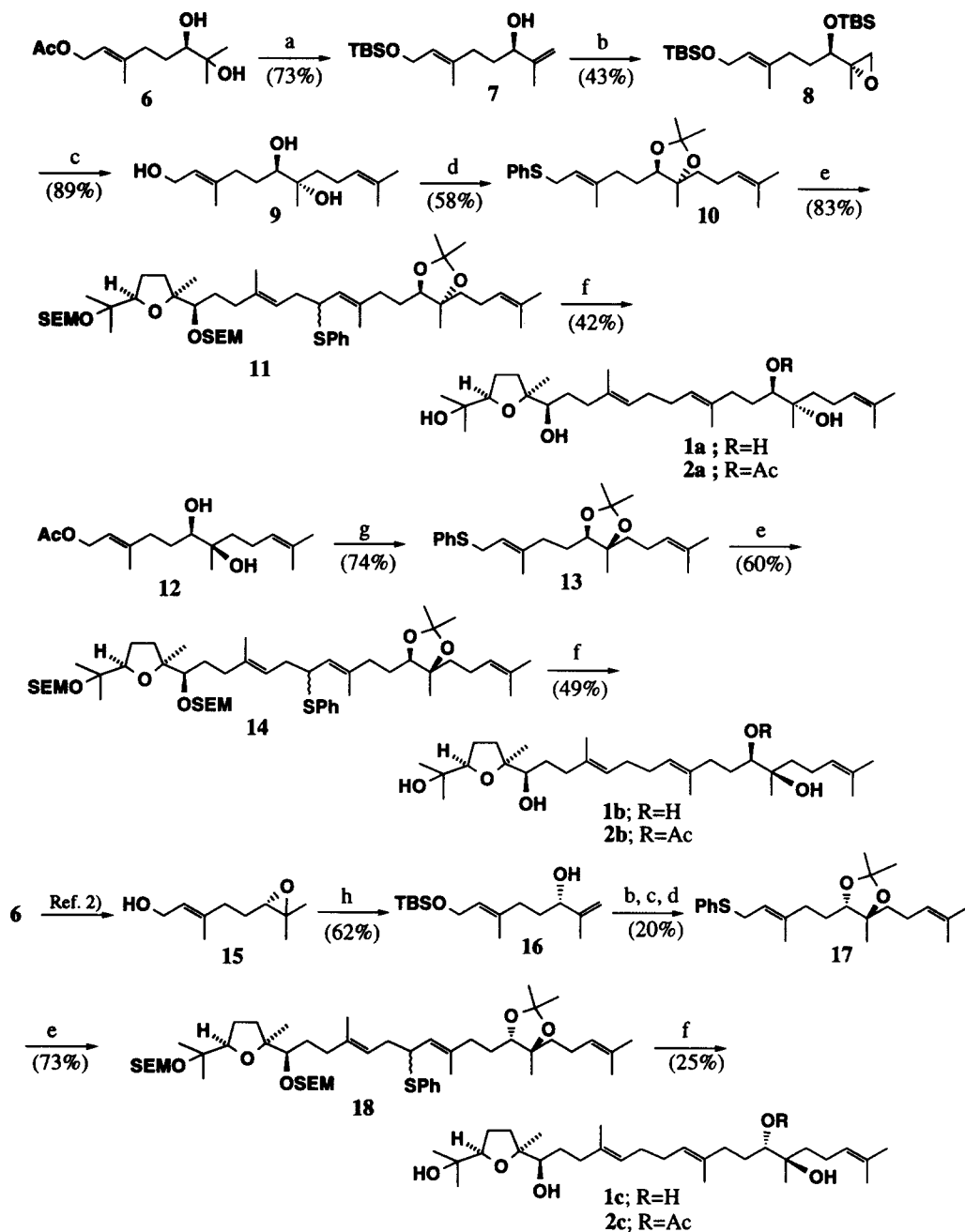
Scheme 2

The right-half segment (6*R*,7*S*)-**B** was synthesized through **D** as follows. The known (*R*)-diol **6**² was transformed into the allylic alcohol **7** by acetylation, dehydration, hydrolysis, and selective silylation of primary alcohol. Epoxidation of **7** using Sharpless' method⁶ took place stereoselectively to give (*S*)-epoxide **8** as the major product. Thus obtained **8** was reacted with prenyl Grignard reagent in the presence of cuprous iodide to afford the triol **9** in high yield after desilylation. The 1,2-diol part of **9** was protected as an acetonide and the terminal hydroxyl group was converted into phenyl sulfide giving **10**.

Reaction of lithio-anion of **10** with **5** afforded the coupling product **11** in high yield. Desulfurization followed by the purification with AgNO₃-impregnated silica-gel chromatography and the hydrolysis of SEM and acetonide protecting groups furnished (3*R*,6*S*,7*R*,18*R*,19*S*)-quassiol A (**1a**) whose ¹H and ¹³C NMR spectra were indistinguishable from those of natural quassiol A. The optical rotation of **1a** ([α]_D +8.3°) was very close to that of natural quassiol A ([α]_D -9.8°)¹ with opposite sign. However, comparison of the optical rotation of 18-monoacetate disclosed that synthetic **2a** ([α]_D +8.4°) and natural **2** ([α]_D +4.9°)¹ are not enantiomeric each other.

In order to clarify the relative configuration at C-18 and C-19, (6*R*,7*R*)-**13** was prepared from the known diol **12**⁷ and reacted with the chloride **5** as described above. Similar desulfurization and deprotection of the coupling product **14** afforded (3*R*,6*S*,7*R*,18*R*,19*R*)-quassiol A (**1b**). The ¹H and ¹³C NMR spectra of **1b** were apparently different from those of natural quassiol A,⁸ which revealed that quassiol A has *anti* 1,2-diol moiety at C-18 and C-19.

Finally, we challenged to the synthesis of (3*R*,6*S*,7*R*,18*S*,19*R*)-quassiol A (**1c**). The (*S*)-epoxide **15**, readily available from **6** was converted to an allylic alcohol **16**, which was then transformed into the sulfide **17** by the same procedure described above. Coupling of **17** with the chloride **5** followed by the similar desulfurization and deprotection gave rise to **1c**. Although the ¹H and ¹³C NMR spectra of **1c** was again identical to those of natural product, the optical rotations of **1c** ([α]_D +0.9°) and its 18-acetate **2c** ([α]_D +1.1°) were completely



Scheme 3

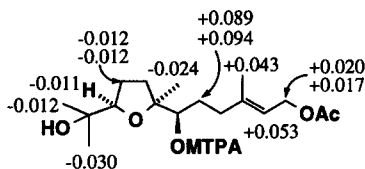
different from those of natural products, **1** and **2**.

Thus, we have synthesized three stereoisomers corresponding to quassiol A and propose the enantiomer of **1a** as the most probable stereostructure of natural quassiol A on the basis of the facts that ^1H and ^{13}C NMR spectral data of **1a** as well as the specific rotation are well consistent with those of natural product.⁹⁾

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References and Notes

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- 5) $\Delta\delta$ ($=\delta_S - \delta_R$) values of MTPA ester of **4a**:



As the following NMR data reveal, quassiol A has (3*R**,6*S**,7*R**)-configuration.

	1, 25 ^{13}C	^1H	2 ^{13}C	3 ^{13}C	^1H	4 ^{13}C	5 ^{13}C	6 ^{13}C	7 ^{13}C	^1H	26 ^{13}C	^1H
4a	25.96	1.11	72.64	85.98	3.80	27.53	33.53	87.17	77.35	3.43	23.01	1.14
	27.01	1.20										
4b	25.21	1.12	72.14	86.65	3.75	27.72	34.88	88.02	76.98	3.38	22.79	1.13
	27.72	1.16										
1	25.93	1.10	72.62	85.87	3.79	27.54	33.29	87.24	77.44	3.44	23.25	1.12
	27.04	1.19										
7-epi 1a	25.90	1.10	72.68	86.33	3.82	27.55	35.83	86.91	77.83	3.35	25.41	1.11
	27.01	1.20										

- 6) Rossiter, B. E.; Verhoeven, T. R.; Sharpless, K. B. *Tetrahedron Lett.* **1979**, 4733-4736.
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- 8) ^1H and ^{13}C NMR data of **1**, **1a**, and **1b** around the diol part (in CD_3OD):

	17 ^{13}C	18 ^{13}C	^1H	19 ^{13}C	20 ^{13}C	29 ^{13}C	^1H
1	30.52	78.13	3.27	75.35	39.37	22.01	1.09
1b	30.75	77.93	3.32	75.58	39.16	22.27	1.10
1a	30.61	78.20	3.28	75.42	39.44	22.05	1.10

- 9) In the private communication, Prof. Tinto, who has isolated quassiols A and B, agreed with the opinion that the stereostructure of quassiol A is the enantiomer of **1a** after our results have been presented. The reason for the inconsistency of optical rotation between **2** and **2a** is not clear.

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